## Answer Key for Checking for Understanding: Using the Periodic Table

1. How many protons does each of the following elements have?

2. What is the chemical symbol for each of the following?

Lithium Li Fluorine $\mathrm{F} \quad$ Sodium Na Mercury Hg Lead Pb
3. What is the name of each of the following?

4. How many electrons does each of the following have?

5. What is the rounded atomic weight (mass number) of each of the following?

6. How many neutrons does each of the following elements have?
P 16
Rb 47
$\begin{array}{lll}U \quad 147 & \mathrm{Cl}\end{array}$ $\qquad$ Bi $\underline{126}$ Pt 117
7. To which family does each of the elements belong?

$$
\begin{array}{llllllllllll}
\mathrm{K} & 1 & \mathrm{C} & 4 & \mathrm{~S} & 6 & \mathrm{Sr} & 2 & \mathrm{Se} & 6 & \mathrm{Ne} & 8
\end{array} \quad \mathrm{Ga} \quad 3
$$

8. In which period is each of the elements located?

9. Which of these would most likely combine in a chemical reaction?
a) $\mathrm{Mg}+\mathrm{O}$
b) $\mathrm{Na}+\mathrm{K}$
c) $\mathrm{S}+\mathrm{Ne}$
d) $\mathrm{O}+\mathrm{Ne}$
